

# Investigation of rare earth distribution in $\text{Sr}_2\text{MgSi}_2\text{O}_7:\text{Eu}^{2+},\text{Dy}^{3+}$ nanophosphors prepared by wet-chemical routes

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Luminescent materials have been developed through fine chemistry methods that help to achieve a better control of parameters such as morphology, particle size, atomic homogeneity and high purity single phase in low temperature synthesis.[1, 2] In this work, the material  $\text{Sr}_{1.98}\text{MgSi}_2\text{O}_7$  nanoparticles doped with 0.01 mol of  $\text{Eu}^{2+}$  and codoped with 0.01 mol of  $\text{Dy}^{3+}$  was prepared via Pechini and Condensation methods. Post-annealing by microwave assisted method using granular coal as the susceptor/reducing agent [3] was applied on both materials and their luminescent properties were compared. The structural position of  $\text{Eu}^{2+}$  used as the activator ions determines photoluminescence properties. The luminescence spectra of  $\text{Sr}_2\text{MgSi}_2\text{O}_7:\text{R}^{n+}$  nanomaterial (R:  $\text{Eu}^{2+}$ ,  $\text{Dy}^{3+}$ ) shows a high emission broad band assigned to the interconfigurational transition  $4f^65d^1 \rightarrow 4f^7$  centered around 460 nm, which is overlapped with a low emission lines attributed to the  $^4F_{9/2} \rightarrow ^6H_{13/2}$  transition of  $\text{Dy}^{3+}$  ion (Fig. 1b). Elemental mappings obtained by Energy Dispersive X-Ray (EDX) presents dopants more likely to be found at the edge, indicating a possible segregation of rare earths to the grain boundaries during the synthesis (Fig. 1c). The persistent luminescence phenomenon emitting in a blue region was observed for both nanomaterials.

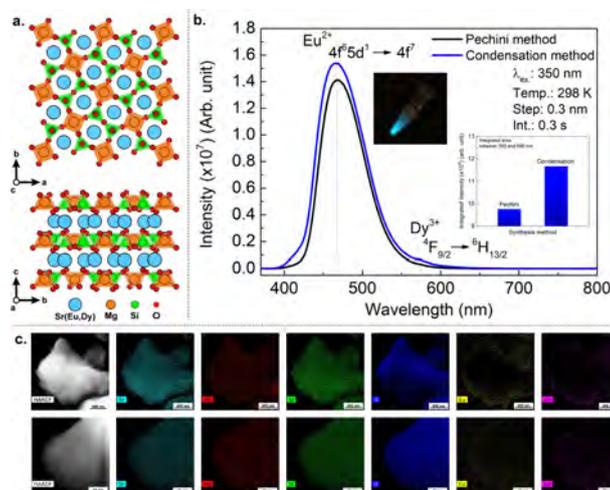


Figure 1: (a) Crystal structure of SMSO view along [001] and [100] axes. (b) emission spectra of both compounds. (c) 2D-elemental mapping of SMSO particles obtained by Pechini (top row) and Condensation (bottom row) methods.

**Keywords:** Persistent luminescence, wet-chemistry, rare earth, nanophosphors, microwave assisted.

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