

Luminescence properties of silica materials incorporating [Eu(ACAC)₃(H₂O)₃] complex and EuCl₃·6H₂O

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Rare earth ions are generally used as fluorescent probes for analytical application because of their distinguished luminescence characteristics such as narrow spectral width, long luminescence lifetime and large Stokes shift. Luminescent functionalized silica particles promising candidates for biotechnological application. In this work, we prepared silica particles containing [Eu(ACAC)₃(H₂O)₃] complex (ACAC = acetylacetonate) and EuCl₃·6H₂O, by modified Stöber method [1]. The emission spectra (Fig. 1) were recorded at room temperature in the range of 420–750 nm, under excitation in the ⁷F₀→⁵L₆ transition. The spectra exhibit narrow bands attributed to ⁵D₀→⁷F_J transitions (where J=0, 1, 2, 3 and 4), dominated by the ⁵D₀→⁷F₂ hypersensitive transition. It is also observed as a broadened peak of the ⁵D₀→⁷F_J transitions in the materials, which can be explained by a non-homogeneity of Eu³⁺-sites due to the silica porous microstructure. However, the intensity of the ⁵D₀→⁷F₂ transition in relation to the ⁵D₀→⁷F₁ transition decreases in the emission spectra for both samples, suggesting that the rare earth ion is in a site with higher symmetry when coated by the silica network. This behavior indicates a network interaction with the Eu³⁺ ion, suggesting a similar chemical environment of the Eu³⁺ ions for both samples. These data suggest that the silica network could interact with the complex or europium chloride, and transfer energy to the central Eu³⁺ ions may be the same way.

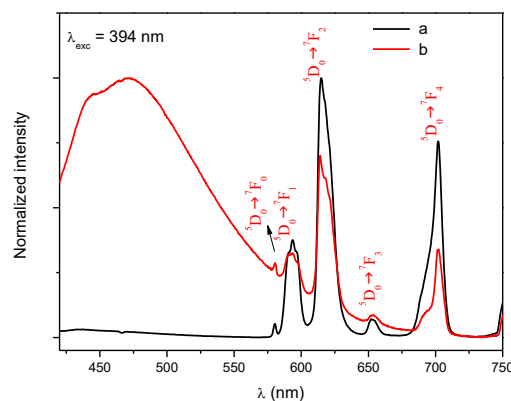


Figure 1 - Emission spectra of silica particles incorporating a) Eu(ACAC)₃(H₂O)₃ complex and b) EuCl₃·6H₂O under excitation at 394 nm.

Keywords: silica, europium, europium chloride, complex, luminescence.

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References

[1] W. Stöber, A. Fink, E. Bohn, *J. Colloid Interface Sci.* 26 (1968) 62–69.